

# CONTRIBUTION OF REFINERY CARBON DIOXIDE EMISSION TO GLOBAL WARMING

By: R. Desrina<sup>\*)</sup>

Researcher at "LEMIGAS" R & D Centre for Oil and Gas Technology  
Jl. Ciledug Raya Kav. 109, Cipulir, Kebayoran Lama, Jakarta Selatan 12230, Indonesia  
Tromol Pos: 6022/KBYB-Jakarta 12120, Telephone: 62-21-7394422, Faxsimile: 62-21-7246150  
First Registered on 28 August 2010; Received after Corection on 20 September 2010;  
Publication Approval on : 30 September 2010

## ABSTRACT

*The energy sector, including petroleum refining, is likely to feature in any legislation aimed at reducing CO<sub>2</sub> emissions. It seems that petroleum refinery contributes relatively small amount of CO<sub>2</sub> emission compared to other sectors such as transportations.*

*Recently, through presidential speech in Copenhagen, government of Indonesia has committed to reduce CO<sub>2</sub> emission to 26 percent in the year of 2020. Many technologies can be used for reducing CO<sub>2</sub> emission in refinery. These technologies include fuel replacement, gasification of heavy residue which leads to single point CO<sub>2</sub> capture, and CO<sub>2</sub> sequestration.*

*This paper tries to discuss how far CO<sub>2</sub> emissions contributed by refinery and possible actions that can be managed on the refinery to significantly reduce CO<sub>2</sub> emissions.*

**Key words:** *global warming, refinery, CO<sub>2</sub> emission, environment.*

## I. INTRODUCTION

Production of CO<sub>2</sub> globally has been brought into discussions in recent years through declarations such as the Kyoto Protocol<sup>(1)</sup>, and also by industries committing to tangible reductions. In essence, the Kyoto Protocol leads a global problem into a national policy, by setting emission targets for greenhouse gases compared with a baseline 1990 level, with a view to reducing global emissions.

How individual nations and industry react to the growing pressures to reduce CO<sub>2</sub> is still to be formulated and ratified. A key element is whether specific industries should be targeted and whether CO<sub>2</sub> trading should be allowed across national boundaries and/or industries.

The energy sector, including petroleum refining, is likely to feature in any legislation aimed at reducing CO<sub>2</sub> emissions. It seems that petroleum refineries contribute relatively small amount of CO<sub>2</sub> emission compared with other sectors such as transportation. However, in this case the refineries may see

real benefits and opportunities in adopting a CO<sub>2</sub> management and reduction strategy. This CO<sub>2</sub> reduction strategy is not only to get benefit from some of the economic gains of CO<sub>2</sub> reduction, but also in order to be seen as a "good neighbour". This CO<sub>2</sub> reduction strategy, for example, through energy conservation and CO<sub>2</sub> utilization, and applying technology. Such technologies, among others, are gasification, which allow heavy residue destruction, relatively easy CO<sub>2</sub> capture and other environmental benefits.

Recently, through presidential speech in Copenhagen, government of Indonesia has committed to reduce CO<sub>2</sub> emission to 26 percent in the year of 2020. This paper tries to discuss how far CO<sub>2</sub> emissions contributed by refineries and possible actions that can be managed on the refineries to significantly reduce CO<sub>2</sub> emissions.

## II. SOURCES OF REFINERY CO<sub>2</sub> EMISSIONS

The petroleum refining industry converts crude oil into thousands of refined products, including lique-

---

<sup>\*)</sup> Lecturer in Petroleum Engineering, FKTE the University of Trisakti

fied petroleum gas, gasoline, kerosene, aviation fuel, diesel fuel, fuel oils, lubricating oils, and feed stocks for the petrochemical industry. Petroleum refinery activities start with receipt of crude for storage at the refinery, followed by all petroleum handling and refining operations, and they terminate with storage preparatory to shipping the refined products from the refinery.

The petroleum refining industry employs a wide variety of processes. A refinery's processing flow scheme is largely determined by the composition of the crude oil feedstock and the chosen petroleum products.

All emissions from the refinery originate from the feed stocks that are used. These feed stocks are the main crude oil(s) to be processed, and other imported feed stocks such as condensates or vacuum gas oils (VGOs), and supplementary natural gas for fuel or hydrogen plants. Carbon is found in the petroleum products that are produced, such as gasoline, diesel oil, etc., with the significant loses emitted into the environment. While most carbon emissions from the refinery will be in the form of CO<sub>2</sub>, there are also other emissions, such as volatile organic compounds (VOCs), coke on catalysts and other minor emissions. Coke on catalyst is not really air borne emission, since it could be land filled. "Shadow" emissions may be

from energy outside of the refinery. This could be CO<sub>2</sub> emissions derived from production of energy offsite. These CO<sub>2</sub> emissions are not emitted from the refinery itself, but are still important when considering the impact of the refinery operations.

Whilst there are many sources of air pollution (e.g. sulfur dioxide and fugitive emission) in the refinery, it seems that two processes in refinery have great contribution to CO<sub>2</sub> emission. These are hydrocracking and fluidized-bed catalytic cracking (FCC) processes<sup>(2)</sup> (see Figure 1).

CO<sub>2</sub> emissions are dominated by those resulting from burning of fuel in fired heaters (approximately 50%) and in utility boilers (approximately 20%). In practice, the refinery will have a large number of process heaters scattered around the site. This makes "end pipe" solution for CO<sub>2</sub> capture is difficult, extremely expensive and even impractical. However, there is potential for capture of the CO<sub>2</sub> produced from power generation, hydrogen production and utilities, which represents approximately half the refinery CO<sub>2</sub> emission.

### III. CO<sub>2</sub> REDUCTION

Literature search indicates that refinery contributions of CO<sub>2</sub> emission are very small comparing to those other activities utilizing petroleum (see Figure 2).

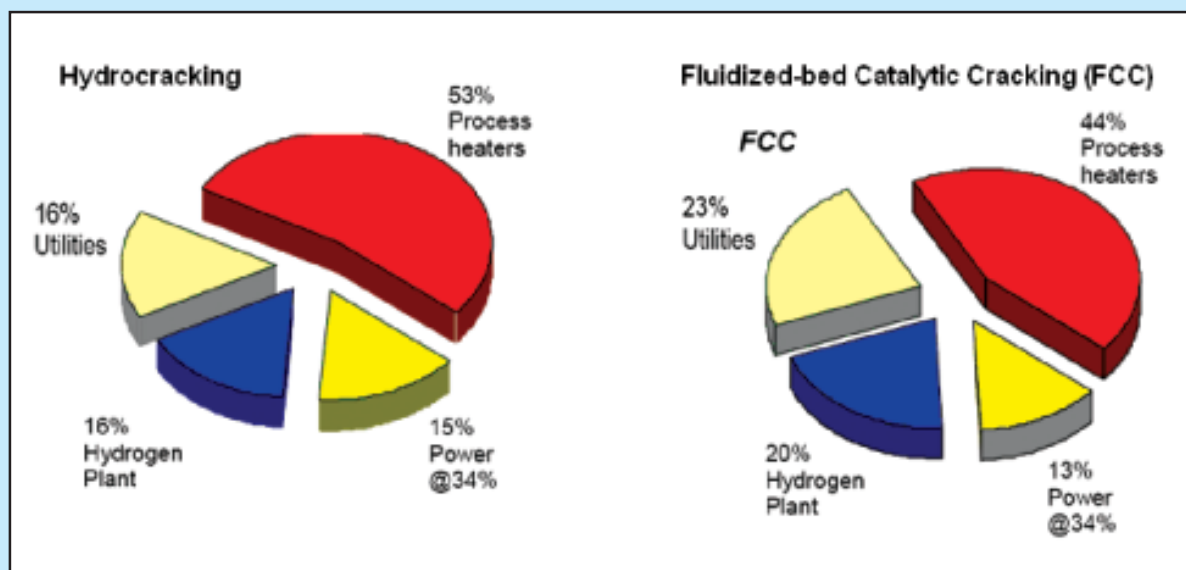


Figure 1  
Sources of refinery CO<sub>2</sub> emission

Figure 2 shows that approximately 90% of CO<sub>2</sub> emissions are derived from combustion of the final product (e.g., utilization of petroleum for energies, such as power plant and vehicles). The refining activity itself is only contributing around 5%. This 5% of CO<sub>2</sub> emissions in the refinery is dominated by process heater which contributes around 45 to 50% (see Figure 1). CO<sub>2</sub> emissions from the refinery might not be measured exactly, since it depends on the final products and the variety process that are utilized<sup>(3)</sup>. Besides, a significant amount of CO<sub>2</sub> is emitted in flares and incinerator which is poorly measured. As approximation, model such as CORINAIR can be used that suggests utilizing factor of 3.14 kg of CO<sub>2</sub> per ton of refinery feed<sup>(4)</sup>.

Although, a relatively small contributor to CO<sub>2</sub> emissions in the oil sector overall, the refinery can be expected to come under pressure to reduce emissions. Legislators appear to group the refining activity with the power generation sector and other stationary sources. The key to understanding the emissions from the refinery is to understand the carbon balance of the refinery.

There are two issues which influence the refinery CO<sub>2</sub> production, i.e:

- Fuel replacement
- The need for hydrogen

In recent years many refineries have replaced some refinery fuel from heavy, high sulfur fuel oil towards refinery fuel gas/natural gas. Actually, the driving force for this switch has been SO<sub>2</sub> reduction. Regardless of operational costing, replacement from heavy fuel oil to natural gas has an impact on CO<sub>2</sub> emissions, e.g., from approximately 0.20 tone to 0.25 tone CO<sub>2</sub> per Mton fuel oil (20% reduction). The use of hydrogen-rich fuel can reduce CO<sub>2</sub> significantly (to approximately 0.075 tone CO<sub>2</sub> per Mton fuel oil). But replacement with hydrogen-rich fuel has to be considered for cost effectiveness and in some cases is difficult to implement on an existing refinery.

The demand for hydrogen on the refinery continues to grow because of the specifications for sulfur content in transportation fuels that consequently results in increased hydroprocessing of products. It also because of processing of higher sulfur crudes lead to increase the need for hydroprocessing.

The demand for hydrogen in the refinery has also been continued when there are introduction of resi-

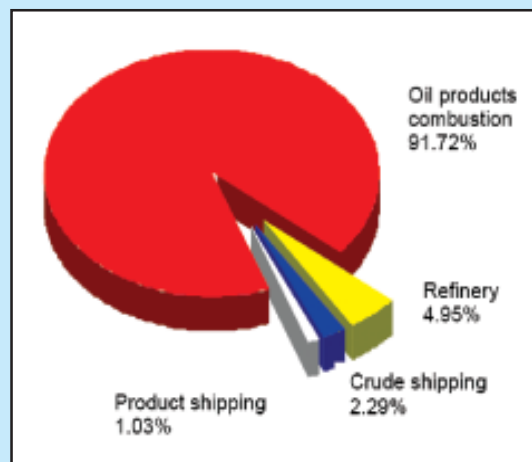


Figure 2  
Typical breakdown of CO<sub>2</sub> emissions  
in the oil industry

due upgrading technologies which require hydrogen in the primary conversion process. Hydrogen is also used for product stabilization and sulfur reduction.

The need for hydrogen has lead to many refineries for recognizing the value of hydrogen and the need to optimize its use through hydrogen analysis. However, often additional hydrogen production is unavoidable. This will be resulting in a significant increase in CO<sub>2</sub> on the refinery. This CO<sub>2</sub> may need to be captured or CO<sub>2</sub> reductions have to be compensated.

Approximately 10 tons of CO<sub>2</sub> per ton of hydrogen are produced. This figure is higher compared to other processes such as steam methane reforming of natural gas or gasification of heavy residues. However, the CO<sub>2</sub> produced in the hydrogen production processes can be captured relatively easily and be sequestered or utilized.

In addition, gasification of residues can be used to provide utilities and hydrogen production, whilst at the same time allowing a single point source for CO<sub>2</sub> capture, such as amine treating process<sup>(5)</sup>. Gasification is a clean, highly efficient to power generation and hydrogen production. Gasification can allow the refineries to decrease production of heavy fuel oil, which is one of the major generators of CO<sub>2</sub>. Moreover, through the development gas-to-liquids (GTL) technologies, synthetic gas (syngas) that is produced could be used to selectively produce diesel oil, the transportation fuel that is much in demand.

Another technique for CO<sub>2</sub> reduction is CO<sub>2</sub> sequestration<sup>(6)</sup>. CO<sub>2</sub> sequestration for enhanced oil recovery (EOR) offers an exciting opportunity for both upstream and downstream oil businesses. Enhanced oil recovery is able to extend the useful life of oil fields increasing production significantly. Many refineries are located on or nearby coasts and many are near operating oil fields.

The upstream sector would clearly benefit from additional oil output. This benefit, for examples, includes a share in upstream revenues from the incremental oil production from EOR, and CO<sub>2</sub> trading across national boundaries.

### III. CONCLUSIONS

Although CO<sub>2</sub> emission from refineries is relatively small (e.g. for approximation 3.14 kg of CO<sub>2</sub> per ton of refinery feed), refineries are actively emitting CO<sub>2</sub> and giving CO<sub>2</sub> impact of their operations and it is expected that refineries will come under increasing pressure to reduce or capture CO<sub>2</sub>.

“End of pipe” solutions for CO<sub>2</sub> reduction from process heater are expensive and on most refineries impractical to implement.

On typical FCC and hydrocracker based refineries gasification of low value refinery residuals could be used to raise utilities and hydrogen, allowing the relatively easy capture of the CO<sub>2</sub> emissions on the refinery.

Another solution for capturing CO<sub>2</sub> could have value as an upstream enhanced oil recovery material sequestered. CO<sub>2</sub> capture and utilization in this way may also provide CO<sub>2</sub> trading benefits to refineries.

### V. ACKNOWLEDGEMENT

The author deeply thanks Dr. M. Mulyono, MSc., Retired Senior Researcher of LEMIGAS Oil and Gas Technology Research Center, for his supervision and advices of preparing this paper.

### REFERENCES

1. Anonymous, 1992, Kyoto Protocol to The United Nations Framework Convention on Climate Change, United Nations.
2. Graham Phillips, 2002, Technology Manager Refining, E. Hemisphere Foster Wheeler Energy Limited, Reading, UK, Gasification V, Noordwijk, Holland.
3. Spoor, R.M., 2008, Low Carbon Refinery, Dream or Reality, Hydrocarbon Processing, pp.113-117.
4. Martín, J., Lumbreras, J., and Rodríguez, E., 2003, Testing Flare Emission Factors for Flaring in Refineries, Technical University of Madrid (UPM). Jose Gutierrez Abascal, 2. 28006. MADRID. .
5. Amrmstrong, T. and Gardner, A. 1998, Amine Treating, Refining Detailed Note Book, Today's Refinery, TPA Inc, Dallas, Texas, USA..
6. Burruss, R.C. and Brennan, S.T., 2003, Geologic Sequestration of Carbon Dioxide, An Energy Resource Perspective, Fact Sheet 26-03, U.S. Geological Survey, U.S Departement of Interior, U.S.A, URL: <http://pubs.usgs.gov/fs/fs026-03/index.html>.